Antioxidant Treatment With N-Acetylcysteine In Endosulfan Intoxication: Report Of Two Cases

U Koca, C Olguner, H Hepaguslar, S Ozkarde?ler, A Gunerli

Citation


Abstract

Objective: Organochlorine compounds may cause oxidative stress by stimulating the production of reactive oxygen species. This mechanism has been implicated in the immunotoxicity, hepatotoxicity and carcinogenicity of these compounds. Endosulfan is an organochlorine insecticide with potential toxicity for central nervous, cardiopulmonary system and liver and kidney. Endosulfan induces T-Cell apoptosis, mitochondrial dysfunction, oxidative stress and lipid peroxidation. N-acetylcysteine (NAC) is an antioxidant agent replenishes the intracellular glutathione stores. Thus, we reported the potential benefits of antioxidant treatment with NAC in endosulfan intoxication.

Setting: Multidisciplinary intensive care unit of the University Hospital

Patients: 16- and 17-year old comatose patients with generalized tonic-clonic convulsions, metabolic acidosis, hyperlactatemia, hypocalcemia, elevated liver and muscle enzymes, trombocytopenia, leukocytosis, normal serum pseudocholinesterase activity, tachycardia and hyperthermia.

Interventions: NAC (150 mg kg-1 body weight with 15 min, 50 mg kg-1 body weight with 4 hours, 100 mg kg-1 body weight with 16 hours) intravenously and cholestyramine 16 g day-1 perorally treatment with standard intensive care unit care.

Results: One of the patients survived, the other patient who ingested highest amount of endosulfan in the literature died. Treatment with N-acetylcysteine improved organ functions and reduced logistic organ dysfunction scores in two cases.

Conclusion: We observed rapid reduction in organ dysfunction parameters and logistic organ dysfunction scores in our two cases suggeting that the potential benefits of NAC as an antioxidant agent in endosulfan intoxication.

INTRODUCTION

Organochlorine compounds are fat-soluble chemical compounds and separated in six groups; dichlorodiphenyltrichloroethene, hexachlorocyclohexane, chlorecone, toxaphene, dicofol and cyclodienes. They are stored in the adipose tissue and resistant to degradation (12).

Endosulfan (C9H6Cl6O3S) is a highly toxic organochlorine belongs to cyclodienes group and devoleped in 1954 under the name of Thiodan (1). It has been shown that endosulfan induces T-cell apoptosis, mitochondrial dysfunction and oxidative stress (13,14) and lipid peroxidation (15).

The cases with acute endosulfan intoxication, accidentally or suicidally, have been reported in literature. The use of the therapeutic modalities such as gastric lavage, activated charcoal, mechanical respiratory assistance, and the agents such as benzodiazepines and barbiturates for seizures have also been reported in endosulfan intoxications (17,18).

N-acetylcysteine (NAC) is a thiol, which is a precursor of L-cysteine and reduced glutathione. NAC is a source of sulphhydryl groups in cells and scaveneger of free radicals as it interacts with reactive oxygen species (ROS) such as hydrogen peroxide and hydroxyl radical (16). It has been shown that administration of NAC prevents liver damage and significantly reduces mortality in acetaminophen overdose (11).

Besides the traditional agents and therapeutic modalities we additionally used NAC in two cases of endosulfan intoxication, because of the therotical benefits of NAC to protect the liver from oxidative injuries.

In our extensive literature search we could find any data
about the use of NAC in the treatment of endosulfan intoxication. We aimed to report two cases with endosulfan intoxication to discuss the potential benefits of NAC as an antioxidant agent and to review the literature of endosulfan intoxication.

**CASE REPORT**

**CASE 1**

A previously healthy 16-year-old man was admitted to a hospital for suicidal ingestion of approximately 100 ml (514 mg kg⁻¹ body weight) of Hektionex 36 EC (endosulfan 360 g L⁻¹) at 07:00 AM. His vital and laboratory findings obtained from history were as follows; On admission at 08:00 AM, the patient was unconscious and had generalized tonic-clonic convulsions. Glasgow coma scale was 3 out of 15, pupils were fix-dilated and nonreactive to light, blood pressure (BP) was 120/70 mm Hg, hearth rate (HR) was 170 bpm. Laboratory tests were arterial pH: 6.95, PₐCO₂: 8.38 kPa, serum bicarbonate: 14 mmol L⁻¹, blood leukocyte cell count: 31.10⁹ L⁻¹, blood hemoglobin: 13.8 g dl⁻¹, blood platelet count 16.8 10⁹ L⁻¹, blood glucose: 3.94 mmol L⁻¹, blood urea nitrogen: 6.39 mmol L⁻¹, serum creatinine: 0.141 µmol L⁻¹, serum alanine aminotransferase: 26.17 µkat L⁻¹, serum aspartate aminotransferase: 25.12 µkat L⁻¹, serum creatine kinase (CK): 25.12 µkat L⁻¹, serum potassium: 4.5 mmol L⁻¹.

He has been treated with 270 mEq NaHCO₃, 10 mg h⁻¹ IV artopine infusion, 1 g pralidoxime and gastric lavage with activated charcoal performed for suspected anticholinesterase poisoning. Serum cholinesterase activity was normal at 122.38 µkat L⁻¹ (normal range: 73.35-225 µkat L⁻¹). After the blood sampling for laboratory investigations, cardiopulmonary resuscitation (CPR), was performed 12 times with 5-10 minutes because of cardiac arrests. Blood gases analysis after CPR were pH: 7.17, PₐCO₂: 4.12 kPa, PₐO₂: 34.71 kPa, HCO₃⁻: 11 mmol L⁻¹. A cranial computed tomography scan revealed diffuse cerebral edema. At 12:00 PM his blood pressure was 44/27 mm Hg and epinephrine 0.1 mg was given and dopamine 5 g kg⁻¹ min⁻¹ and dobutamine 5 g kg⁻¹ min⁻¹ was started. His serum cholinesterase activity was normal at 122.7 µkat L⁻¹. At 00:30 AM he was transported to intensive care unit (ICU), with his blood pressure 140/70 mm Hg, hearth rate 109 bpm.

On admission to ICU, we started dopamine infusion (15 g kg⁻¹ min⁻¹) at fourth day of admission to ICU and epinephrine (0.03-0.05 g kg⁻¹ min⁻¹) at seventh day of admission to ICU for refractory hypotension despite of optimal fluid resuscitation. At 150 h of admission to ICU brain death tests were started, but he died at the 155 h.

**CASE 2**

A previously healthy 17-year-old man was admitted to a hospital for suicidal ingestion of approximately 100 ml (514 mg kg⁻¹ body weight) of Hektionex 36 EC (endosulfan 360 g L⁻¹) at 10:40 PM. On admission to our university hospital emergency service, his Glasgow coma scale 3 out of 15, pupils were fix-dilated and unresponsive to light. On mechanical ventilation (pressure regulated volum control, FiO₂: 0.7) blood gases analysis were pH: 7.17, PₐCO₂: 4.12 kPa, PₐO₂: 34.71 kPa, HCO₃⁻: 11 mmol L⁻¹. He was transported to our university hospital for mechanical ventilatory assistance. On admission at 11:00 PM to our university hospital emergency service, his Glasgow coma scale 3 out of 15, pupils were fix-dilated and unresponsive to light. On mechanical ventilation (pressure regulated volum control, FiO₂: 0.7) blood gases analysis were pH: 7.48, PₐCO₂: 2.58 kPa, PₐO₂: 34.71 kPa, HCO₃⁻: 14.7 mmol L⁻¹. A cranial computed tomography scan revealed diffuse cerebral edema. At 12:00 PM his blood pressure was 44/27 mm Hg and epinephrine

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**Figure 1**

Table 1: Laboratory and clinical findings of first case

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Normal range/liter (µkat L⁻¹)</th>
<th>24 h</th>
<th>48 h</th>
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emergency service at 02:00 AM, his Glasgow coma scale was 3 out of 15, pupils were fix-dilated and responsive to light, he had generalized tonic-clonic convulsions. Endotracheal intubation and cardiac defibrilation by 200 J were performed for ventricular fibrillation. CPR was performed with 3 minutes. Diazepam 10 mg and thiopentone 200 mg were given to control convulsions. On admission to ICU at 03:00 AM, his blood pressure was 185/115 mm Hg, heath rate was 170 bpm, body temperature was 36 C. Glasgow coma scale 3 out of 15, pupils were fix-dilated and unresponsive to light and he had generalized tonic-clonic convulsions. Blood gases analysis were pH: 7.08, \( P_{a}CO_{2} \): 5.65 kPa, \( P_{a}O_{2} \): 30.38 kPa, HCO\(_{3}^{-}\): 12.8 mmol L\(^{-1}\), oxygen saturation: 0.99 and anion gap were 32 mmol L\(^{-1}\). And then we treated high anion gap metabolic acidosis with 200 mEq NaHCO\(_{3}\). Laboratory and clinical findings of the case 2 during the ICU treatment were outlined in table 2. A computed tomography scan of the cerebrum was normal. We controlled convulsions by infusion of thiopentone infusion 75-200 mg h\(^{-1}\) with 54 hours. Mechanical ventilation, activated charcoal was started. At the 14\(^{th}\) hour of admission to ICU, we started NAC treatment (150 mg kg\(^{-1}\) body weight with 15 min, 50 mg kg\(^{-1}\) body weight with 4 hours, 100 mg kg\(^{-1}\) weight with 16 hours) and cholestyramine 16 g day\(^{-1}\) perorally. At the 7\(^{th}\) hour of admission to ICU, we infused dopamine 15 g kg\(^{-1}\) min\(^{-1}\) with 9 hours for refractory hypotension to optimal fluid resuscitation. He regained consciousness within 80 hours of admission to ICU and was extubated after 7 hours. At 7 day after admission to ICU he was discharged to the medical ward.

**Figure 2**

Table 2: Laboratory and clinical findings of second case

<table>
<thead>
<tr>
<th></th>
<th>1 day</th>
<th>2 day</th>
<th>3 day</th>
<th>4 day</th>
<th>5 day</th>
<th>6 day</th>
<th>7 day</th>
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**DISCUSSION**

Accidental lethal intoxication with endosulfan was first described in humans in 1974 by Terziev et al (13). The reported signs in humans after endosulfan intoxication previously in literature include: abdominal pain, nausea, womiting, foamy oral fluid, chest pain, sinus tachycardia, hypertension (13), miosis (which maybe followed by mydriasis), hyperglycemia, elevated activity of liver enzymes (1,7,8), hyperthermia (1,7,14), rhabdomyolisis (9), acute interstitial nephritis (15), acute tubular necrosis (1), metabolic acidosis combined with a elevated anion gap (1,7), hyperlactatemia (7), respiratory dystress, respiratory arrest, tremor, ataxia, convulsions, unconsciousness (1,13), normal level of serum pseudocholinesterase activity (1), leukocytosis (13), and thrombocytopenia (1). In addition, we experienced elevated serum lactate dehydrogenase (LDH), CK and CK-MB activity and troponin –T and decreased serum ionized calcium level not reported in previous cases. At cellular level, organochlorines may inhibit enzyme activities of the mitochondrial electron transport chain (13). Imbealut et al (13) reported that increased plasma organochlorine levels were significantly corelated with reduced oxidative enzyme activities of skeletal muscle in man. Organochlorine related rhabdomyolisis has been reported previously in human (13,18). Elevated serum CK and LDH levels was probably related to the muscle injury in our cases. Severe rhabdomyolisis in our first case, as judged by highly elevated serum CK, LDH, and myoglobin may be related to prolonged anoxia or toxicity.
of extremely high amount of ingested endosulfan. Rapid
decrease of muscle enzymes in our cases may be attributed
to antioxidant effects of NAC treatment. Although elevated
serum CK-MB activity and troponin-T levels in our first
case may be attributed to myocardial injury by CPR, we
didn’t experience any ischemic electrocardiographic changes
in our cases.

Garg et al. (19) demonstrated that the high dose of endosulfan
(40 mg kg$^{-1}$) significantly decreases the plasma calcium
level with 35% at 4 h after ingestion in rats. We also
experienced decreases in plasma levels of ionized calcium in
our cases.

Organochlorine compounds have the ability to cause
oxidative stress by stimulating the production of ROS, and
this mechanism has been implicated in the immunotoxicity,
hepatotoxicity and carcinogenicity of these chemicals
(20-24). ROS are produced primarily by the mitochondria
in cells as a by-product of normal metabolism during
conversion of molecular oxygen to water. These include:
superoxide radical, hydrogen peroxide, hydroxyl radical.
The mean antioxidative enzymes are catalase which converts
hydrogen peroxide to molecular oxygen and water,
glutathione peroxidase (GPx) which converts hydrogen
peroxide to water, coupled to the oxidation of reduced
glutathione to oxidized glutathione, and superoxide
dismutase (SOD) which converts superoxide radical to
hydrogen peroxide ($\text{H}_2\text{O}_2$). Endosulfan reduced erythrocyte,
lung and liver SOD activity and lung and liver glutathione
(GSH) levels in rat (4). In adrenocortical cells of rainbow
tROUT (OnCorhynchus mykiss), endosulfan reduced GPx
activity and GSH level and increased the activity of
glutathione transferase, also increased lipid hyperperoxides
levels (4). In rats, single dose of 30% LD$_{50}$ endosulfan
increased thiobarbituric acid reactive substance (TBARS) in
cerebral and hepatic tissues (4). The other organochlorine
insecticide chlor dane increased serum skeletal muscle
related creatinine phosphokinase and lactate dehydrogenase
activities and lipid peroxidation in rat liver (25). Sahoo et al
(25) demonstrated that Hexachlorocyclohexane (HCH) is an
organochlorine compound that induces oxidative stress in rat
cerebral hemisphere, as judged by decreased cerebral
catalase, SOD, GPx activity and GSH levels.

Serum liver function tests elevated in our two cases.
Extremely high serum liver enzyme levels in first case may
be due to ingestion of extremely high amount endosulfan
with or without prolonged anoxia. Boerboom et al. (1)
reported that congestion of liver with peri venular steatosis
presumptively attributed to endosulfan intoxication at
forensic autopsy. Elevations of serum transaminases in our
second case identical to previously reported cases in
literature (26). Venkateswarlu et al. (27) reported that the
values of elevated serum transaminases returned to normal at
the end of six weeks in forty four endosulfan poisoning. The
organochlorine lindane causes an increase in activity of the
pro-oxidant enzyme NADPH oxidase in isolated neutrophils
(28) and in liver microsomal preparations (21,22). In the liver
this pro-oxidant activity is accompanied by lipid
peroxidation and hepatocellular injury (21,22). Similarly the
other organochlorine dieldrin induces a state of oxidative
stress in liver by causing an increase in the production of
ROS with a concomitant decrease in antioxidant
concentrations and an increase in hepatic DNA synthesis.
Treatment with antioxidants attenuates all of these changes,
suggesting the involvement of oxidant mechanisms in toxic
and tumor-promoting effects of dieldrin (20,28). Rapid
decreases of elevated serum transaminases in our cases may
be due to NAC treatment. NAC, serves as a precursor for
glutathione and, thus, can replenish the intracellular
glutathione stores (29); however, NAC is also able to act as a
direct scavenging agent (30). Thus NAC produce antioxidant
and cytoprotective effects, furthermore, NAC may stimulate
endothelium-derived relaxing factor and improve
microvascular blood flow (30). In septic patients NAC
increases hepatopulsodynamic blood flow (30), decreases
peroxidative stress (30), attenuates the increase in S
transferase and circulating ICAM-1 and VCAM-1 after
reperfusion of the donor liver, indicating possible
cytoprotective effects of NAC (30). We used logistic organ
dysfunction score (LOD), include neurologic, pulmonary,
cardiovascular, renal, haemotologic and hepatic parameters,
for assessing organ functions (31). LOD scores decreased after
NAC treatment in our both cases.

In rats (32) and cats (33) cerebrum had higher endosulfan
concentration than cerebellum and remaining parts of brain.
The neurotoxicity of endosulfan has been attributed to
inhibition of the calmodulin-dependent Ca$^{2+}$ -ATPase
activity (34), alterations in the serotoninergic system (35)
and inhibited GABA receptors (36). Endosulfan is able to inhibit
sodium-, potassium-, and magnesium-dependent ATPase
enzymes in rainbow trout brain (37). In cats, endosulfan
produced high amplitude bursts of spike discharges in
electrocorticogram. Changes in electrical activity on cerebral
cortex attributed to the direct effect of the compound as the
Antioxidant Treatment With N-Acetylcysteine In Endosulfan Intoxication: Report Of Two Cases

Concentration was highest in this area. The depletion of acetylcholine without any change in acetylcholinesterase activity might be due to the direct effect of endosulfan on the synthesis of acetylcholine (\(\alpha\)). Single oral administration of endosulfan (15 mg kg\(^{-1}\) body weight) to rats reduced phosphatidylinositol and phosphatidylinositol 4,5-bisphosphate without any significant effect on phosphatidylinositol 4-bisphosphate, suggest that possible phosphoinositide messenger system in the neurotoxicity of endosulfan (\(\alpha\)).

In humans convulsions begins 1-4 hours after endosulfan ingestion (\(\alpha\)). In our cases, tonic-clonic generalized convulsions began approximately 1 hour after endosulfan ingestion. Our first and second case admitted to intensive care unit 16\(^{th}\) and 3\(^{rd}\) hour respectively after endosulfan ingestion with Glasgow coma scale of 3 out of 15 and their pupils were fix-dilated and nonreactive to light. We obtained diffuse cerebral edema in CT scan of first case that probably related to repeated cardiopulmonary resuscitation and toxic effect of organochlorine and brain CT scan of second case was normal. Also Haun et al (\(\alpha\)) were obtained diffuse cerebral edema in a fatal toxaphene, another organochlorine, poisoning. In the first case, seizures have been obtained before admission to ICU, but we didn't experience seizures in ICU probably because of severe cerebral damage due to hipoxia and repeated cardiopulmonary resuscitation.

Benzodiazepines and barbiturates can be used for controlling the seizures in endosulfan intoxication. Pentobarbital was an effective therapeutic measure against an absolute lethal dose of endosulfan in rats, while diazepam was not effective (\(\alpha\)). Boereboom et al (\(\alpha\)) reported use of thiopentone (i.v. 10 mg kg\(^{-1}\) body weight bolus over 6 minutes followed by i.v. 20 mg kg\(^{-1}\) body weight over 30 minutes with i.v. maintenance therapy of 5-10 g day\(^{-1}\) ) in endosulfan intoxication. In their case, the first day after admission the EEG showed persistent epileptic activity despite the anticonvulsive herapy. We infused 75-200 mg h\(^{-1}\) thiopentone for 54 hours for controlled seizures in second case.

Our cases were treated with sodium bicarbonate because of metabolic acidosis. We experienced high anion gap metabolic acidosis in the second case and did not in the first case because of treatment with sodium bicarbonate before admission to ICU. Blanco-Coronado et al (\(\alpha\)) reported high anion-gap metabolic acidosis in six patients with endosulfan intoxication. Our patients had severe metabolic acidosis supporting the interference of the agent with the cellular metabolism and oxygen consumption (\(\alpha\)). The rapid correction of metabolic acidosis suggest however that anoxia due to the convulsions may be the primary cause of metabolic acidosis. Also we obtained hyperlactatemia in our cases that identical with previously reported case (\(\alpha\)) that suggests altered cellular metabolism.

Endosulfan can be absorbed following ingestion, inhalation and skin contact. The LD\(_{50}\) of endosulfan varied widely depending on the route of administration, species, vehicle and sex of the animal. The oral LD\(_{50}\) of endosulfan in rats ranges from 18-355 mg kg\(^{-1}\) body weight (\(\alpha\)). Blanco-Coronado et al (\(\alpha\)), reported that six patients with acute endosulfan intoxication. One of these patients whom blood endosulfan level was 2,85 mg L\(^{-1}\) (mean= 0,48 mg L\(^{-1}\) ) had died. Boereboom et al (\(\alpha\)) reported a case that nonaccidental ingestion of endosulfan 260 mg kg\(^{-1}\) body weight had died. Our first case ingested 514 mg kg\(^{-1}\) body weight endosulfan and this is the highest amount of endosulfan ingestion in literature. Male rats given a single oral dose of endosulfan at 40 mg kg\(^{-1}\) body weight displayed neurotoxic manifestations and showed a significant increase in blood glucose, blood ascorbic acid and blood and brain glutathione and decrease in plasma calcium levels (\(\alpha\)). Blanco-Coronado et al (\(\alpha\)) reported hyperglycemia in six case with endosulfan intoxication. We also experienced hyperglycemia in second case and hypocalcemia in both case.

Following acute over-exposure high endosulfan concentrations can temporarily be found in the liver; the concentration in plasma decreases rapidly (\(\alpha\)). The tissue distribution of endosulfan in a case who died with 6 hours after ingestion was maximum in liver followed by kidney, urine and blood (\(\alpha\)).

Therapeutic modalities such as carefully performed gastric lavage with tap water followed by salin purgatives, multiple doses of activated charcoal to minimize systemic absorption of the toxin, cholestyramine, nonabsorbable bile acid binding exchange resin for enhance fecal elimination of the toxin by interrupting the biliary-enterohepatic and enteroenteric recirculation, can be used for the treatment. We also used activated charcoal and cholestiramine in our cases.

In conclusion, the severity of organ injury related to endosulfan intoxication may be correlated with the amount of ingested agent. NAC, as an antioxidant agent may be a supplemental treatment strategy of endosulfan intoxication related organ dysfunction. This suggestion needs further controlled experimental investigations.
Antioxidant Treatment With N-Acetylcysteine In Endosulfan Intoxication: Report Of Two Cases

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