Effect Of Storage Time And Temperature On Some Serum Analytes
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Citation

Abstract
Information on the measured concentration of serum analytes during storage of serum samples is often incomplete and sometimes contradictory. The 10 analytes have not studied in this area in healthy subjects. The aim of present study was designed to determine the effect of storage time and temperature on the laboratory results of 10 analytes in sera from apparently healthy adult males in city of Gorgan. We studied the effect of storage temperature and time on the measured concentration of 10 serum analytes (2006). Serum was separated from the clot within 20 min of the collection. The sera were stored at 4±1 degree °C and 23±1 degree °C for 0, 1, 2, 3, 4, 5, 6, 7, 8, 24, 48, and 72 h, then assayed. Glucose, Phosphorus and creatinine were the least stable and the serum should be determined within 48 h at 4±1 degree °C and 24 h at 23±1 degree °C for these analytes. The other analytes were stable for 72 h. Proper storage temperatures and times must be considered for these analytes (glucose, phosphorus and Creatinine) if measurement is not to take place immediately after specimen collection. Beyond this, it is even very useful to check the reliability of technical and instrumental resources that the laboratory will use during the study because molecular alterations of the analytes due to variable storage conditions can cause misleading results.

INTRODUCTION
Laboratory tests are used by clinicians for diagnosis, monitoring, and prognosis in patients with different diseases. A number of factors, primarily preanalytical and analytical or normal biological variations affect the accuracy of test results. Preanalytical factors such sample collection and handling, diet, exercise and drugs can all impact a test result. The key characteristics of any test are its bias and imprecision. Bias is primarily an analytical characteristic, in which reported results differ from the actual value. Imprecision, or lack of reproducibility, is due to both physiological and analytical factors [1]. To detect real pathological changes in patients, the preanalytical and analytical variations must be reduced to acceptable levels at which they cause no impact on clinical interpretation of the results. Observational clinical studies can often be greatly enhanced by the inclusion of biochemical analyses in stored serum samples collected from the population being studied. Biochemical analyses can be used to detect risk factor exposure, to control for confounding, or to measure the effects of bias. In randomized trials, biochemical analyses can be used to monitor the safety and biochemical efficacy of treatment. Standard guidelines for blood sample handling state that plasma or serum should be separated (20-30 min) from cells as soon as possible after clot formation is complete to avoid clot-induced changes in the concentration of serum analytes [1]. Whilst this is necessary for particular analytes, it might be assumed that many blood analytes deteriorate within a matter of hours in unseparated samples kept at ambient temperature. For most routine assays in a clinical laboratory, serum is the sample. The laboratory receives the specimen in the form of whole blood, and then separates the serum from the clot by centrifugation. For clinically useful and reliable test results, the interval between blood collection and serum separation must be controlled. Many investigators have studied related changes in some analytes, but the results are controversial. The 10 analytes have not studied in this area in healthy subjects. For this reason the present study was designed to determine the effect of storage time and temperature on the laboratory results of 10 analytes in sera from apparently healthy adult males in city of Gorgan. In this study we tried to find out the quantitative alterations and the useful length of stored serum in different time and temperature on the laboratory results.

METHODS
The subjects were healthy adult males about 25-50 years old, from a employees of Gorgan Faculty of Medicine (2006). The adult males were instructed to fast overnight until blood
collection was completed. 10 ml of blood was collected from each male (in total 10 adult males) without the use of an anticoagulant. Samples were allowed to clot at room temperature for 20 min. we then centrifuged one sample from each male and analyzed the separated sera of 10 analytes without delay (zero time).

In addition one sample from each of the 10 adult males was stored for 0,1, 2, 3, 4, 5, 6, 7, 8, 24, 48 and 72 hours at each of the following temperatures: In fridge with 4 ± 1 degree °C which was set up especially for this project and in laboratory bath with 23 ± 1 degree °C.

We determined the concentration of Glucose, Urea, Uric Acid, Creatinine, Albumin, Total protein, Total cholesterol, Calcium, Phosphorus and Triglyceride in serum using laboratory kits and spectrophotometry technique (model JENWAY, IOS UV/VIS) in the laboratory of Biochemistry(Faculty of Medicine). The samples showed visible hemolysis was excluded from the study. Data were analyzed using paired t-test for comparison of different storage time and temperature. Results were considered significant when p<0.05.

ASSAY ANALYTE PROCEDURES

The different assay analytes in this study were as follow:

GLUCOSE MEASUREMENT

The enzyme glucose oxidase catalyzes the oxidation of glucose to gluconic acid. Addition of the enzyme peroxidase, phenol and 4- aminoantipyrine results in the formation of a colored compound (quionone) that can be measured and its absorbance was determined at the wavelength of 500 nm.

UREA MEASUREMENT

The enzyme Urease changes the urea to ammoniac and CO2. Addition of the enzyme glutamate dehydrogenase, L-ketoglutarate and NADH results in the formation of a L-glutamate and NAD that can be measured and its absorbance was determined at the wavelength of 340 nm.

URIC ACID MEASUREMENT

The enzyme Uricase catalyzes the oxidation of uric acid to allantoinie. Addition of the enzyme peroxidase, dichloro 2 hydroxybezen sulfonic acid and 4- aminoantipyrine results in the formation of a colored compound (quionone) that can be measured and its absorbance was determined at the wavelength of 510 nm.

CREATININE MEASUREMENT

Creatinine react with picrate ion in an alkaline medium to yield an orange-red complex. Whose absorbance is measured at 510 nm.

ALBUMIN MEASUREMENT

Serum albumin reacts with bromcresol green (BCG) to form a colored compound that can be measured and its absorbance was determined at the wavelength of 630 nm.

TOTAL PROTEIN MEASUREMENT

The biuret method depends on the presence of peptide bonds, which react Cu ++ ions in alkaline solutions to form a colored product whose absorbance is measured at 550 nm.

TOTAL CHOLESTEROL MEASUREMENT

The enzyme cholesterol esterase catalyzes the cholesterol ester to cholesterol and fatty acid. Addition of the enzyme cholesterol oxidase results in the formation of 4-cholesten-3-one and H2O2. Addition of the enzyme peroxidase, phenol and 4- aminoantipyrine results in the formation of a colored compound (quionone) that can be measured and its absorbance was determined at the wavelength of 500 nm.

CALCIUM MEASUREMENT

Colorimetric measurement, with methylmol blue. that can be measured and its absorbance was determined at the wavelength of 612 nm.

PHOSPHOROUS MEASUREMENT

One of the used methods for serum inorganic phosphate are based on the reaction of phosphate ions with ammonium molybdate to form a phosphomolybdate complex.

TRIGLYCERIDE MEASUREMENT

The enzyme lipase catalyzes the triglycerides to glycerol and fatty acid. Addition of the enzyme glycerokinase results in the formation of glycerol-3-phosphate. Addition of the enzyme glycerol-3-phosphate oxidase results in the formation of dihydroxyacetone phosphate and H2O2. Addition of the enzyme peroxidase, N-ethyl-1-N-sulfopropy 1 N-anisidine and 4- aminoantipyrine results in the formation of a colored compound (quionone) that can be measured and its absorbance was determined at the wavelength of 546 nm.

RESULTS

The effects of storage time and temperature on results for the following assays were not statistically significant: Urea, Uric
Acid, Calcium, Total protein, Albumin, Triglyceride and Total cholesterol. Statistically significant changes as compared with the initial time values (table 1) were noticed for 3 constituent (table 2 and 3).

**Figure 1**

Table 1: Assay values in sera separated immediately after collection (Zero time) from 10 healthy males (with reference intervals).

<table>
<thead>
<tr>
<th>Analyte</th>
<th>Mean</th>
<th>Standard Deviation</th>
<th>Reference intervals</th>
</tr>
</thead>
<tbody>
<tr>
<td>Glucose (mg/dl)</td>
<td>84.7±0.56</td>
<td></td>
<td>70-105</td>
</tr>
<tr>
<td>Urea (mg/dl)</td>
<td>34.5±0.81</td>
<td></td>
<td>15-45</td>
</tr>
<tr>
<td>Uric Acid (mg/dl)</td>
<td>4.6±0.27</td>
<td></td>
<td>3-7</td>
</tr>
<tr>
<td>Creatinine (mg/dl)</td>
<td>0.8±0.03</td>
<td></td>
<td>0.6-1.3</td>
</tr>
<tr>
<td>Total cholesterol (mg/dl)</td>
<td>179±0.73</td>
<td></td>
<td>150-260</td>
</tr>
<tr>
<td>Triglyceride (mg/dl)</td>
<td>93.7±2.75</td>
<td></td>
<td>60-105</td>
</tr>
<tr>
<td>Total protein (g/dl)</td>
<td>7.6±0.12</td>
<td></td>
<td>6-8.3</td>
</tr>
<tr>
<td>Albumin (g/dl)</td>
<td>4.2±0.38</td>
<td></td>
<td>3.5-5.2</td>
</tr>
<tr>
<td>Calcium (mg/dl)</td>
<td>2.1±0.26</td>
<td></td>
<td>8.0-10.2</td>
</tr>
<tr>
<td>Phosphorus (mg/dl)</td>
<td>3.4±0.29</td>
<td></td>
<td>2.7-4.5</td>
</tr>
</tbody>
</table>

1. Method-specific reference intervals established by commercial kits.

2. mg/dl (milligram/deciliter), g/dl (gram/deciliter).

**Figure 2**

Table 2: Assay values changes in analyte concentration over time in sera stored at 4±1 degree °C.

**Figure 3**

Table 3: Assay values changes in analyte concentration over time in sera stored at 23±1 degree °C.

**Figure 4**

Table 3: Assay values changes in analyte concentration over time in sera stored at 23±1 degree °C.

**Figure 5**

Table 3: Assay values changes in analyte concentration over time in sera stored at 23±1 degree °C.
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For urea, Uric Acid, Calcium, Total protein, Albumin, Triglyceride and Total cholesterol there was no effect on samples stored at 4 ± 1 degree °C and 23 ± 1 degree °C for as long as 72 h (table 2 and 3). Tests that were suitable for analysis at 48 h incubation at 4± 1 degree °C but not suitable for analysis at 48 h incubation at 23±1 degree °C (suitable for analysis at 24h) were glucose, phosphorus and creatinine (table 2 and 3).

DISCUSSION

There is a lack of consensus regarding the most appropriate specimen type for analysis of many biochemistry analytes. Information on the stability of serum analytes during storage of serum is often incomplete and sometimes contradictory. In this study, effect of storage at room temperature(23±1 degree °C) and refrigeration (4±1 degree °C) for 0, 1, 2, 3, 4, 5, 6, 7, 8, 24, 48 and 72 hours on 10 sera analyses were investigated. In this study we found that of the 10 sera analytes that we measured, only glucose, creatinine and phosphorus were affected by storage at 4 ±1 degree °C and 23 ±1 degree °C for 48 h and 24 h respectively. Glucose was decreased after 48 h and 24 h at 4 ±1 degree °C and 23 ±1 degree °C respectively. Creatinine and phosphorus were increased after 48 h and 24 h at 4±1 degree °C and 23 ±1 degree °C respectively. An important advantage of this study when compared with those already published was the number of different storage times.

Donnelly et al [13] investigated the stability of 25 analytes from serum of healthy donors and stored at room temperature and 4 degree °C over 48 h, 14 days and 4 months respectively. All 10 analytes were stable at 2 temperature for specified times. Study of Saeed et al [14] on camel serum showed that albumin, calcium, Phosphorus and cholesterol did not change over 9 days when stored at 4-5 degree °C. At 4-5 degree °C, creatinine and glucose in camel serum remained stable for 6 days, Total protein for 7 days and blood urea nitrogen for 8 days. At room temperature (23-25 degree °C). Total protein, albumin, calcium and phosphorus were stable throughout 9 days. Changes in glucose occurred after 3 days. Bobby et al [15] investigated the stability of 24 analytes after immediate separation of serum and stored at room temperature(25 degree °C) and analyzed in 0, 2, 4, 8, 16, 24, 32, 40, 48 and 56 h after collection. All analytes in serum were stable over 56 h periods. Heins et al [16] determined the effects of storage time and temperature on 22 serum analytes. In serum at room temperature, phosphorus, uric acid and triacylglycerols increased continuously. Our results for serum analytes were consistent with those obtained by previous studies [13-16] who investigated these serum analytes. But the results of this study were not in agreement with the results of other investigators [13]. Changes in the concentration of Glucose, phosphorus and creatinine clinically significant with increasing storage temperature. At temperature 23 ±1 degree °C the concentration of glucose in serum decreased and the concentration of phosphorus and creatinine in serum increased with increasing temperature when compared with initial results and 4±1 degree °C storage temperature. The decrease of glucose concentration during storage may be related to sensitivity of glucose to temperature or distinguishing of glucose in higher temperaure. It has been suggested that the increase in creatinine concentration during storage is due to non-specific formation of pseudocreatinines [16] with kinetic jaffe reaction. An abnormally high phosphorus and creatinine concentration could be also result of prolonged storage of serum at room temperature (23 ±1 degree °C). Recording the length of time from collection to separation of each sample might allow appropriate adjustment to be made for the slight increase in concentration of these analytes over time. It should also be noted that room temperature in the present study was 23±1 degree °C which may not be realistic for studies in hotter climates. Some routine tests can tolerate fairly long delays (72 h) in room temperature without changes in analyte content. Samples for glucose, phosphorus and creatinine should be processed within 24 h and 48 h. The remaining analytes evaluated were stable for 72 h. Proper storage temperature and times must be considered for these analytes (Glucose, Phosphates and Creatinine) if measurement is not to take place immediately after specimen collection. Beyond this, it is even very useful to check the reliability of technical and instrumental resources that the laboratory will use during the study because molecular alterations of the analytes due to variable storage conditions can cause misleading results. In conclusion we hope that the results we have presented will help assess which of the constituents may be assayed in serum stored for prolonged times under commonly encountered storage conditions when such prolonged storage occurs in advertently or is unavoidable. We recommend that Samples should analyse in the laboratory within preferably 24 h of collection to ensure valid results. In addition, the turn-around time from sample drawing to reporting the analytical result would be shortened.
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